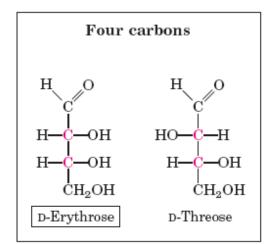
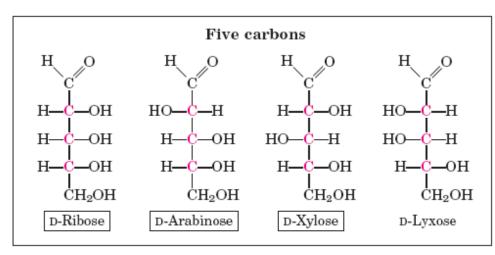
# Carbohydrates

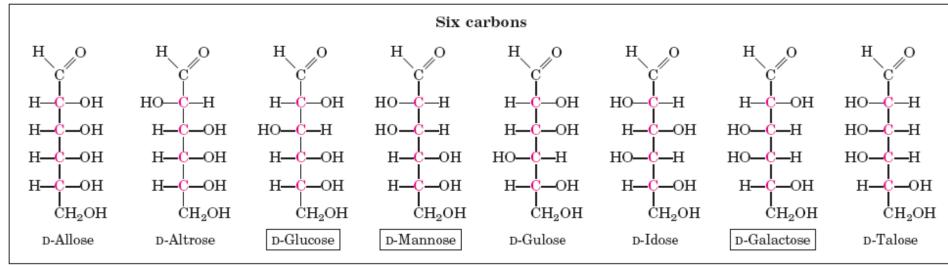
Three carbons

H
O
C
H
$$-$$
C
 $-$ 
CH
 $_2$ 
OH

D-Glyceraldehyde





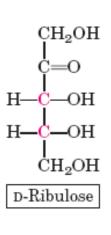


**D-Aldoses** 

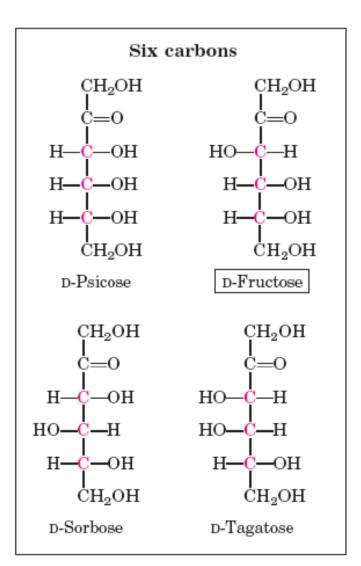




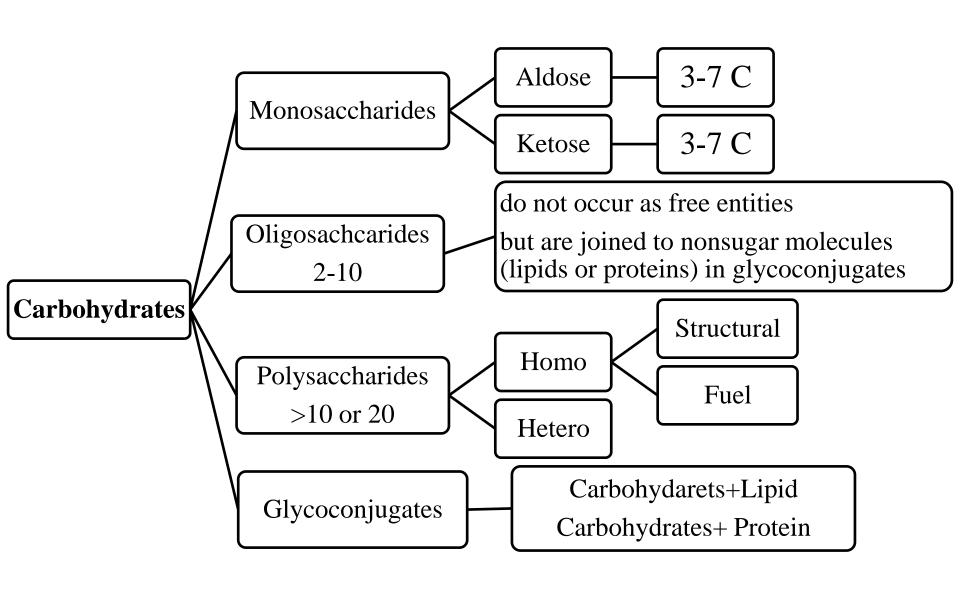
$$\begin{array}{c} \operatorname{CH_2OH} \\ | \\ \operatorname{C=O} \\ | \\ \operatorname{H-C-OH} \\ | \\ \operatorname{CH_2OH} \\ \\ \operatorname{D-Erythrulose} \end{array}$$



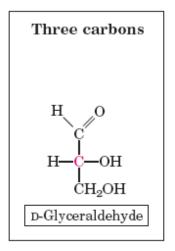
$$CH_2OH$$
 $C=O$ 
 $HO-C-H$ 
 $H-C-OH$ 
 $CH_2OH$ 
 $D-Xylulose$ 

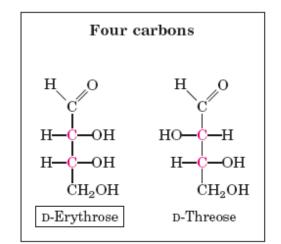


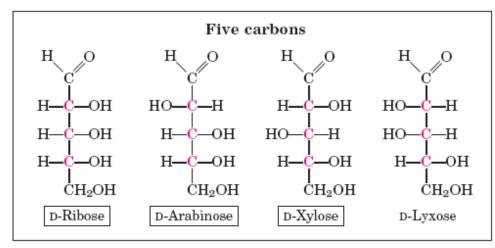
#### **D-Ketoses**

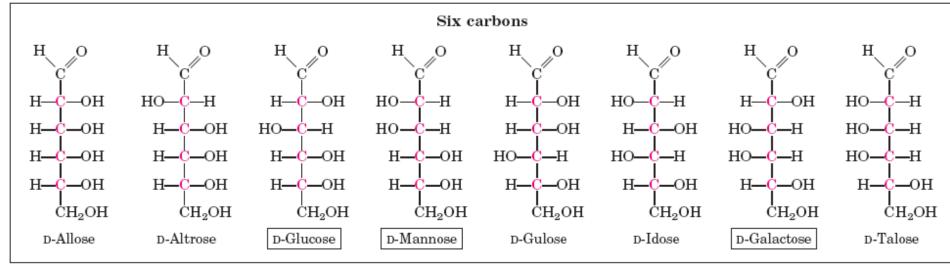


# مونوساكاريدها داراي مراكز نامتقارن مي باشند.



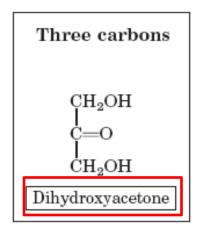


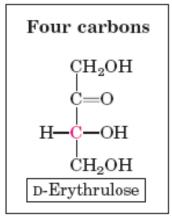




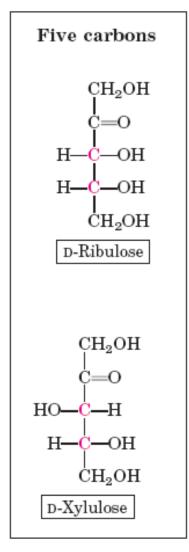
#### D-Aldoses

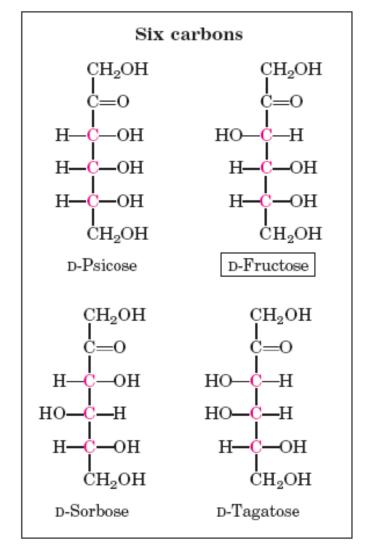
# مونوساکاریدها دارای مراکز نامتقارن می باشند.



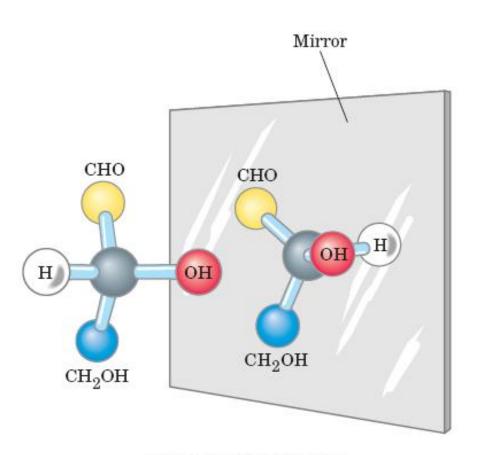


All the monosaccharides except dihydroxyacetone contain one or more asymmetric (chiral) carbon atoms and thus occur in optically active isomeric forms.

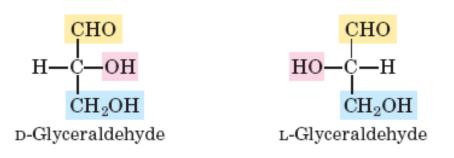




#### **D-Ketoses**



Ball-and-stick models



### Fischer projection formulas

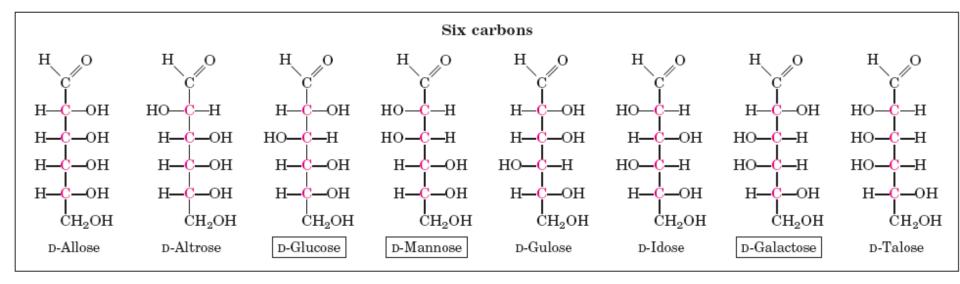


Perspective formulas

## Enantiomer

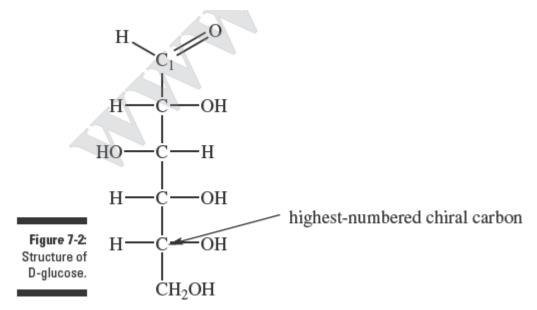
تعداد ایزومرهای فضایی مولکولی با n مرکز کایرال = ???? فرم D و D

 $2^{4}=16$   $2^{n}=1$ مرکز کایرال  $2^{n}=16$ تعداد ایزومرهای فضایی مولکولی با

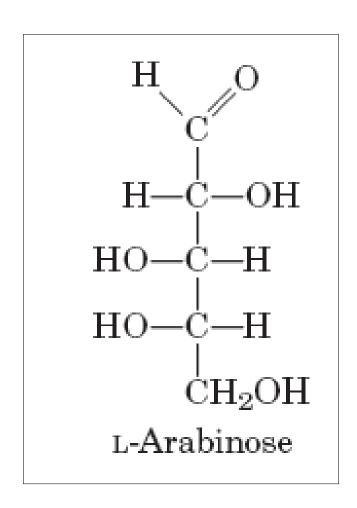


D-Aldoses

تعیین ایزومر D و L در مونوساکاریدهای با بیش از یک مرکز کایرال:



بیشتر قندهای موجود در طبیعت از نوع ایزومر D هستند.



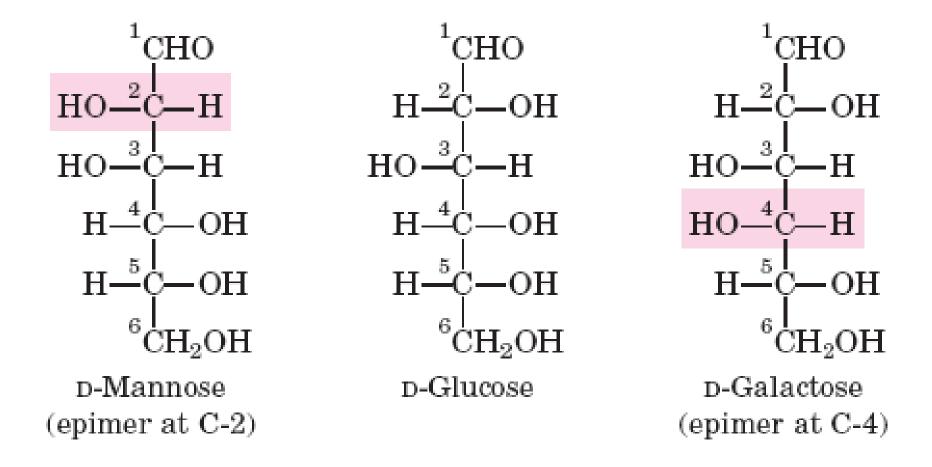
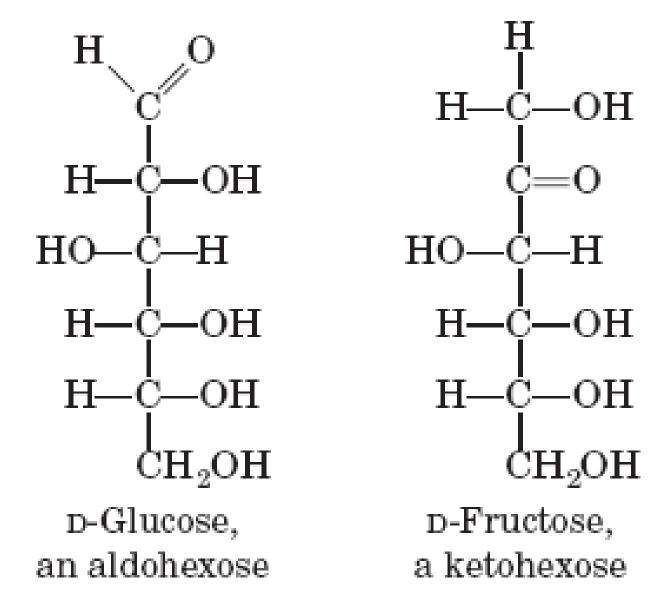
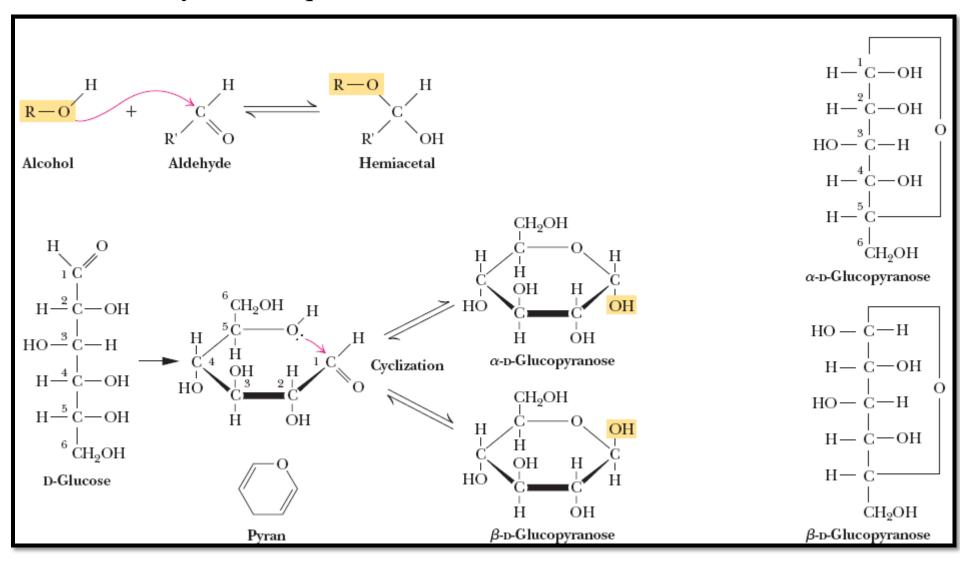


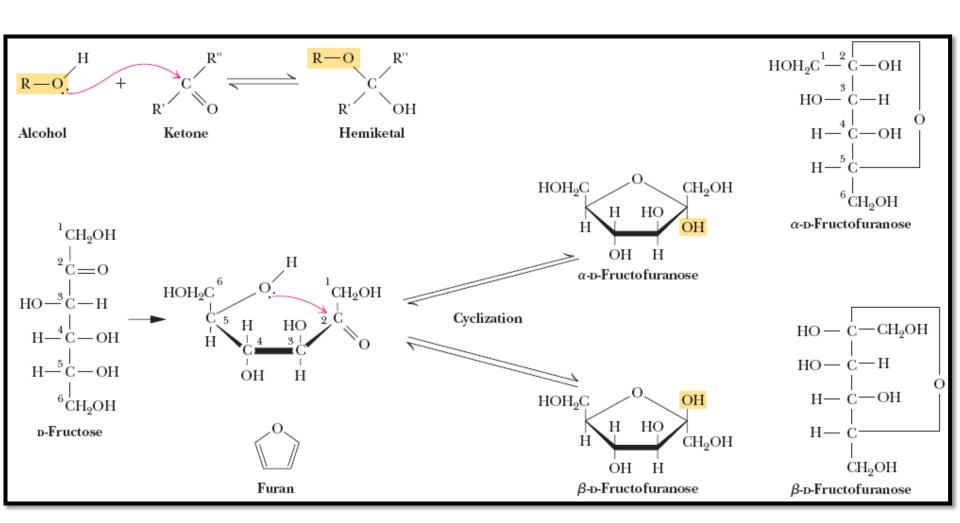
FIGURE 7-4 Epimers. D-Glucose and two of its epimers are shown as projection formulas. Each epimer differs from D-glucose in the configuration at one chiral center (shaded red).

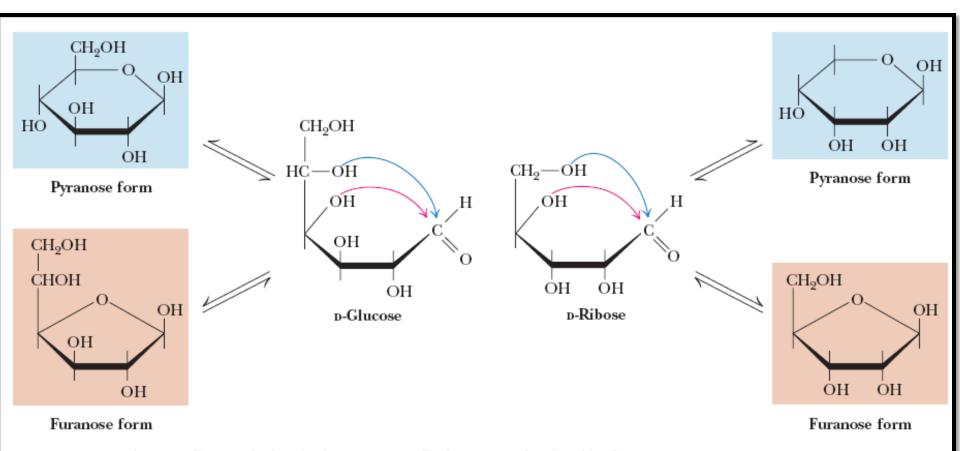
# The Common Monosaccharides Have Cyclic Structures



# > 4 C has cyclic in aques solution







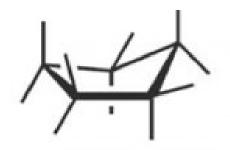
**FIGURE 7.7** D-Glucose, D-ribose, and other simple sugars can cyclize in two ways, forming either furanose or pyranose structures.



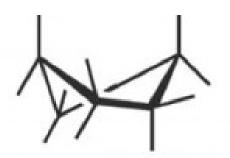
$$\begin{array}{c} H \\ HO \\ HO \\ H \end{array} \begin{array}{c} CH_2OH \\ OH \\ H \end{array} \begin{array}{c} OH \\ OH \\ OH \end{array} \begin{array}{c} OH \\ OH \\ OH \end{array}$$

**(b)** 

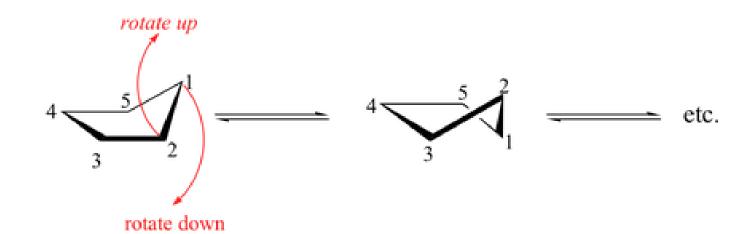
**FIGURE 7.8** (a) Chair and boat conformations of a pyranose sugar. (b) Two possible chair conformations of  $\beta$ -D-glucose.

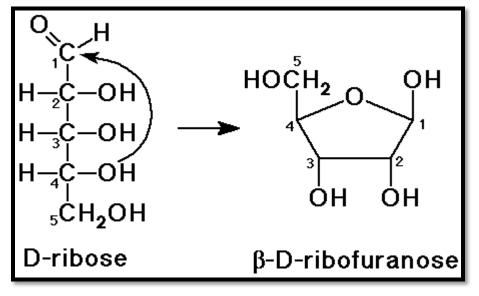


Envelope



Twist

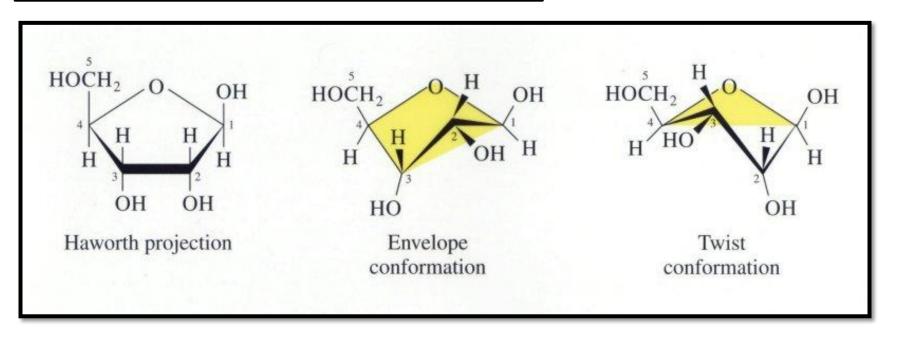


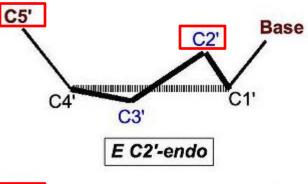


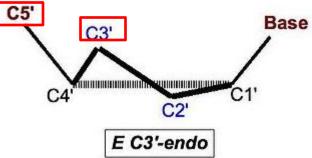
In solution, the straightchain (aldehyde) and ring ( $\beta$ -D-furanose) forms of **free** ribose are in <u>equilibrium</u>.

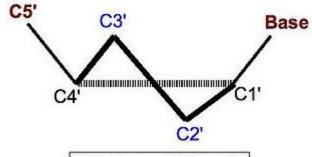
Envelope: only a single atom is displaced

Twists: Two atoms is displaced

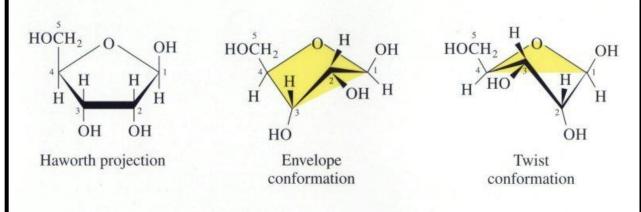


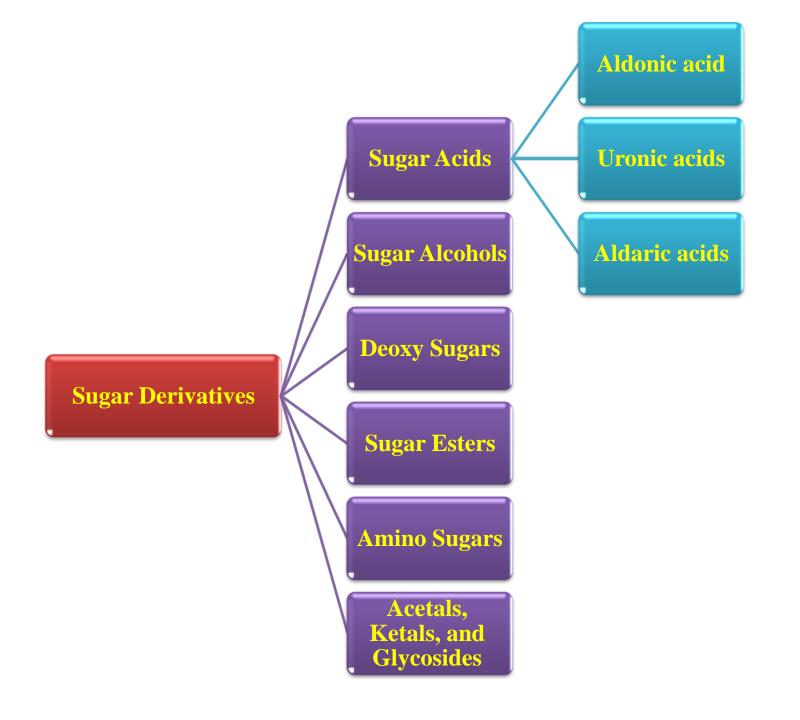






T C2'-exo-C3'-endo





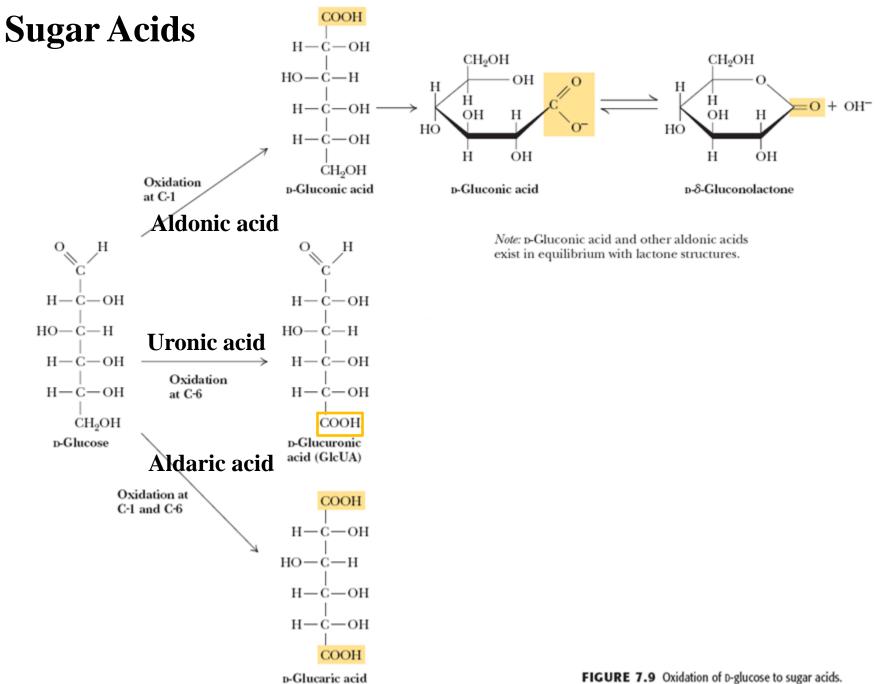
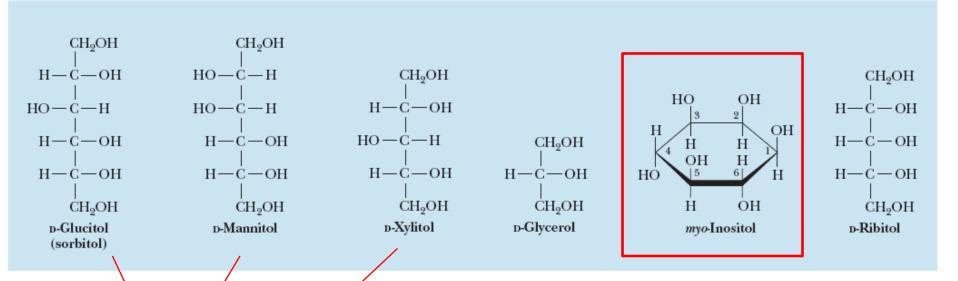


FIGURE 7.9 Oxidation of D-glucose to sugar acids.

addition of alkaline CuSO<sub>4</sub> (called *Fehling's solution*) to an aldose sugar produces a red cuprous oxide (Cu<sub>2</sub>O) precipitate:

O O 
$$\parallel$$
 RC—H + 2 Cu<sup>2+</sup> + 5 OH<sup>-</sup>  $\longrightarrow$  RC—O<sup>-</sup> + Cu<sub>2</sub>O $\downarrow$  + 3 H<sub>2</sub>O Aldehyde Carboxylate



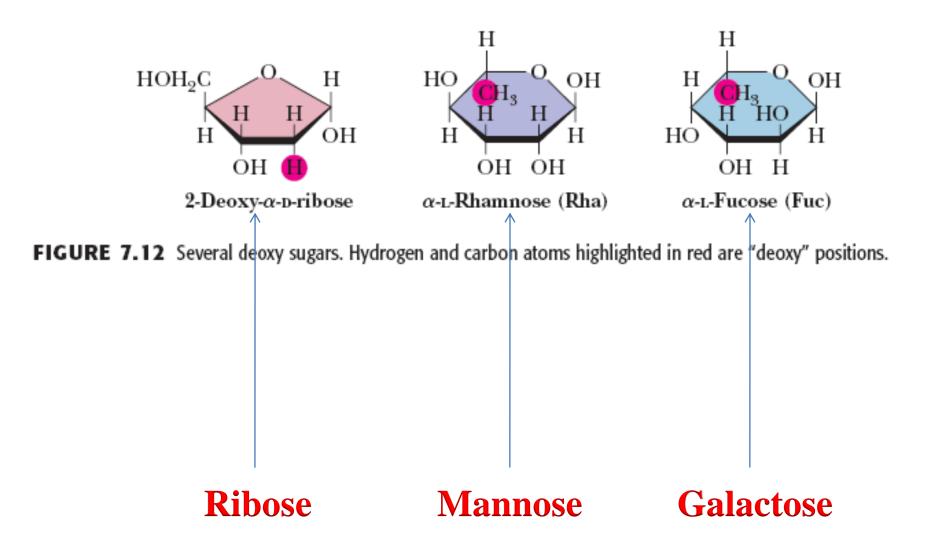
**FIGURE 7.10** Structures of some sugar alcohols. (Note that *pryo*-inositol is a polyhydroxy cyclohexane, not a sugar alcohol.)



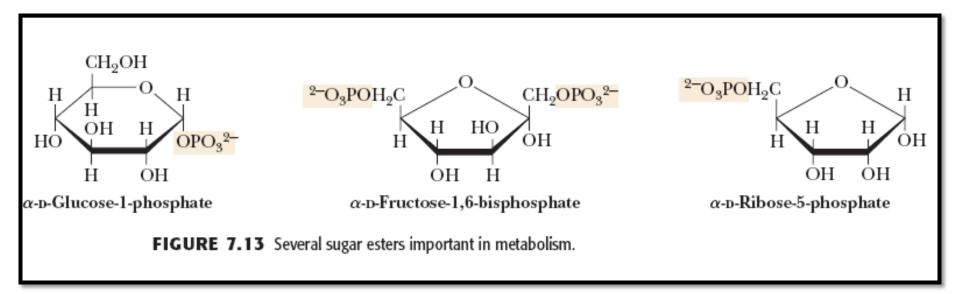
FIGURE 7.11 Sugar alcohols such as sorbitol, mannitol, and xylitol sweeten many "sugarless" gums and candies.

# **Sugar Alcohols**

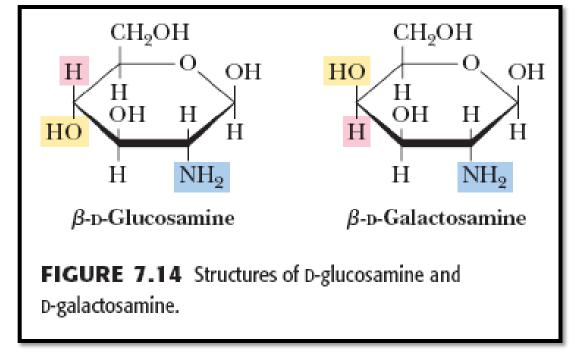
# **Deoxy Sugars**

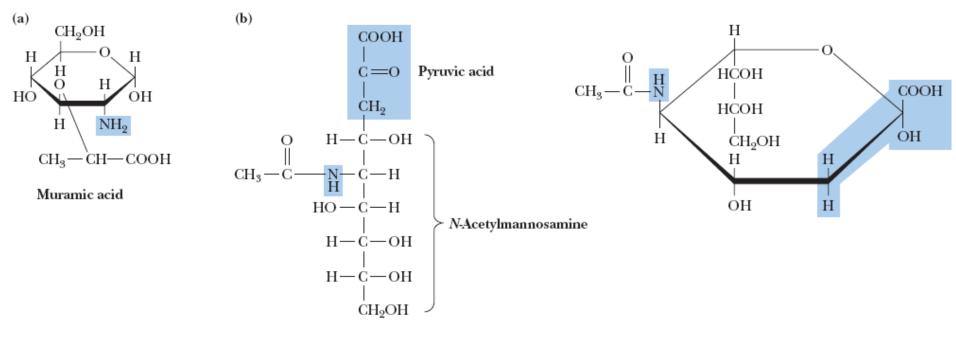


### **Sugar esters**



## **Amino Sugars**





N-Acetyl-p-neuraminic acid (NeuNac, also known as Neu5Ac), a sialic acid

FIGURE 7.15 Structures of (a) muramic acid and (b) several depictions of a sialic acid.

FIGURE 7.16 Acetals and ketals can be formed from hemiacetals and hemiketals, respectively.

## **Acetal, Ketals and Glycosides**

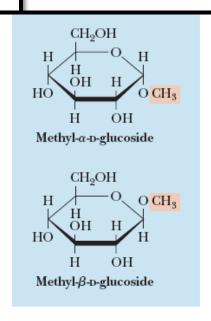
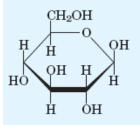
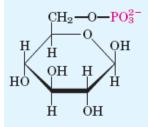


FIGURE 7.17 The anomeric forms of methyl-D-glucoside

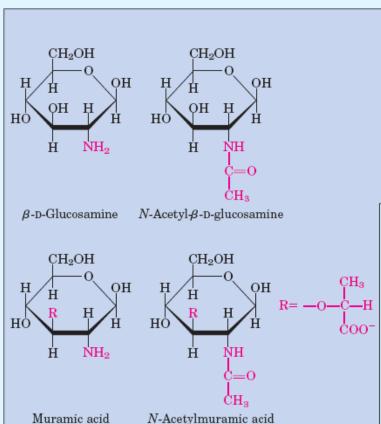
#### Glucose family



 $\beta$ -D-Glucose



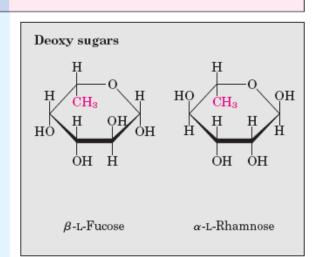
β-D-Glucose 6-phosphate

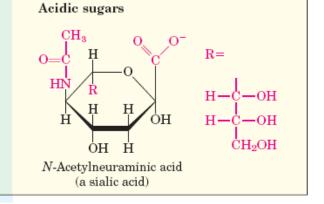


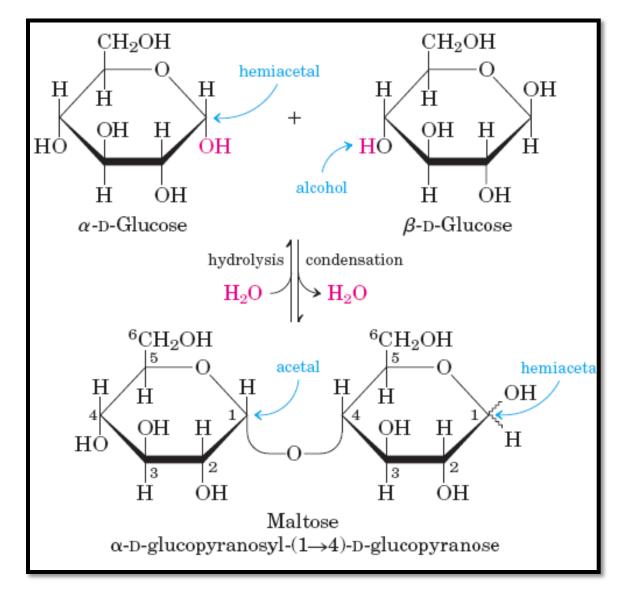
## 

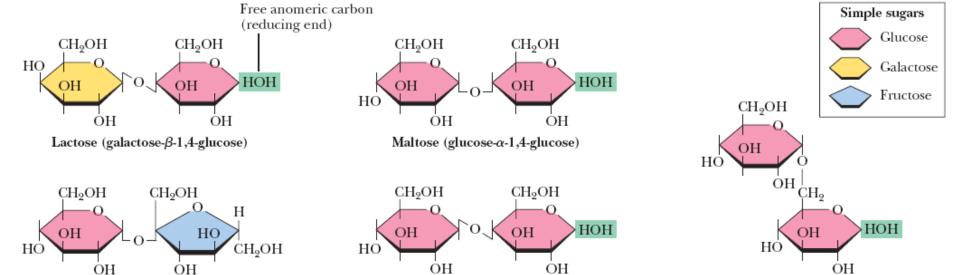
 $\beta$ -D-Mannosamine

 $\beta$ -D-Galactosamine





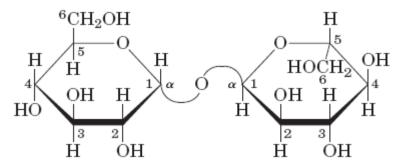




Cellobiose (glucose-β-1,4-glucose)

**FIGURE 7.18** The structures of several important disaccharides. Note that the notation "HOH" means that the configuration can be either  $\alpha$  or  $\beta$ . If the —OH group is above the ring, the configuration is termed  $\beta$ . The configuration is  $\alpha$  if the —OH group is below the ring. Also note that sucrose has no free anomeric carbon atom.

Isomaltose (glucose- $\alpha$ -1,6-glucose)



Sucrose (glucose-1α-β2-fructose)

Trehalose  $\alpha$ -D-glucopyranosyl  $\alpha$ -D-glucopyranoside  $\operatorname{Glc}(\alpha 1 \leftrightarrow 1\alpha)\operatorname{Glc}$ 

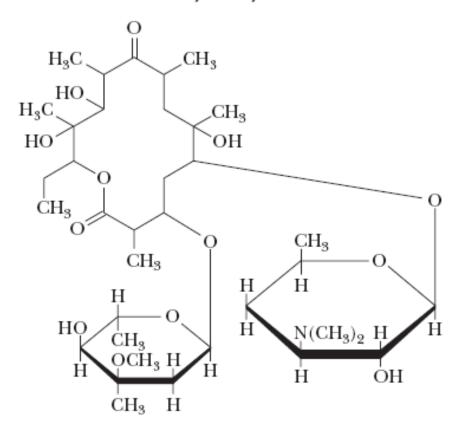
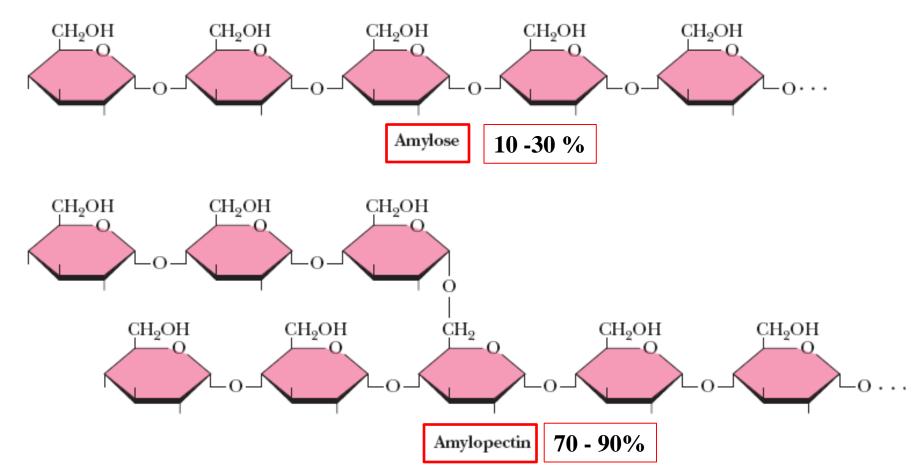


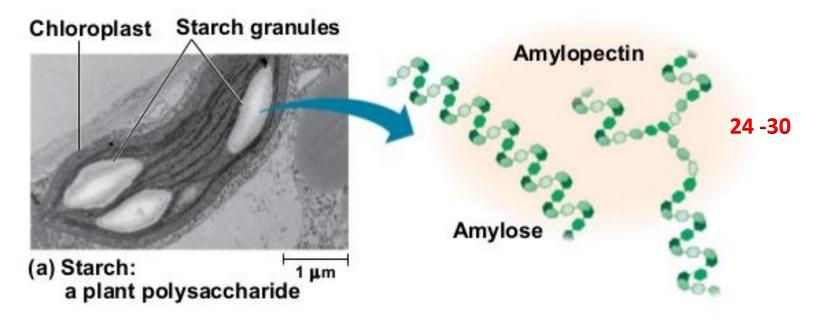
FIGURE 7.19 Some antibiotics are oligosaccharides or contain oligosaccharide groups.

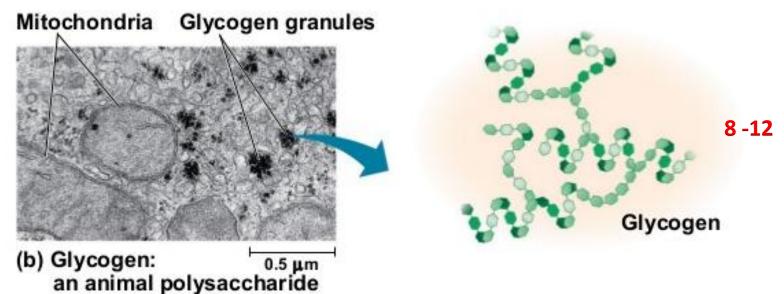
# Homopolysaccharides Heteropolysaccharides Unbranched Branched Two Multiple monomer monomer types, types, unbranched branched

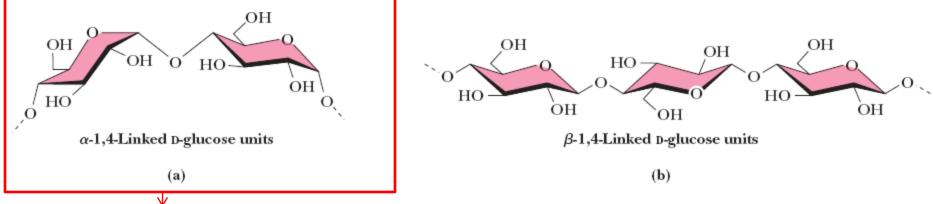
**FIGURE 7-13 Homo-** and heteropolysaccharides. Polysaccharides may be composed of one, two, or several different monosaccharides, in straight or branched chains of varying length.



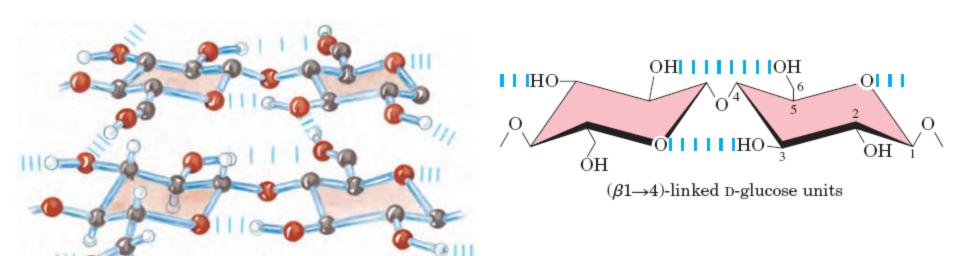
**FIGURE 7.20** Amylose and amylopectin are the two forms of starch. Note that the linear linkages are  $\alpha(1\longrightarrow 4)$  but the branches in amylopectin are  $\alpha(1\longrightarrow 6)$ . Branches in polysaccharides can involve any of the hydroxyl groups on the monosaccharide components. Amylopectin is a highly branched structure, with branches occurring every 12 to 30 residues.

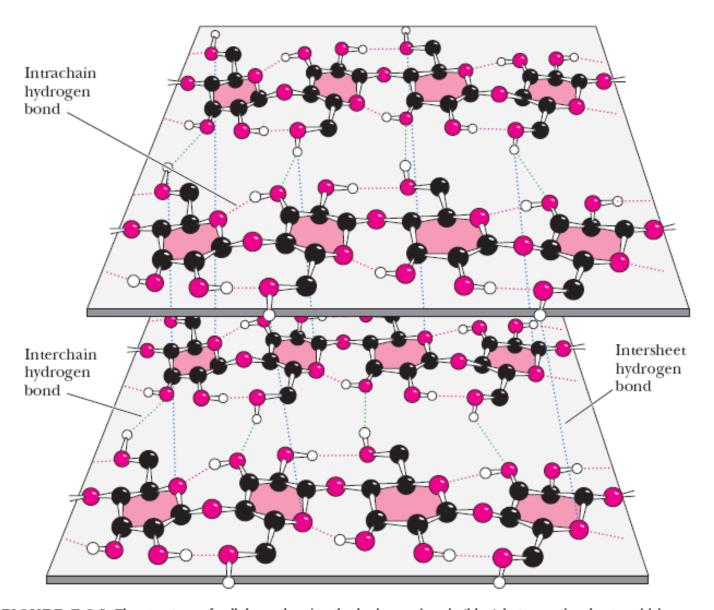






**FIGURE 7.23** (a) Amylose, composed exclusively of the relatively bent  $\alpha(1 \longrightarrow 4)$  linkages, prefers to adopt a helical conformation, whereas (b) cellulose, with  $\beta(1 \longrightarrow 4)$ -glycosidic linkages, can adopt a fully extended conformation with alternating 180° flips of the glucose units. The hydrogen bonding inherent in such extended structures is responsible for the great strength of tree trunks and other cellulose-based materials.





**FIGURE 7.24** The structure of cellulose, showing the hydrogen bonds (blue) between the sheets, which strengthen the structure. Intrachain hydrogen bonds are in red, and interchain hydrogen bonds are in green. (Illustration: Irving Geis. Rights owned by Howard Hughes Medical Institute. Not to be reproduced without permission.)

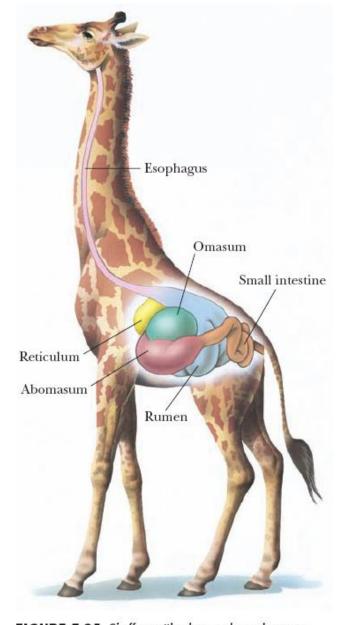


FIGURE 7.25 Giraffes, cattle, deer, and camels are ruminant animals that are able to metabolize cellulose, thanks to bacterial cellulase in the rumen, a large first compartment in the stomach of a ruminant.

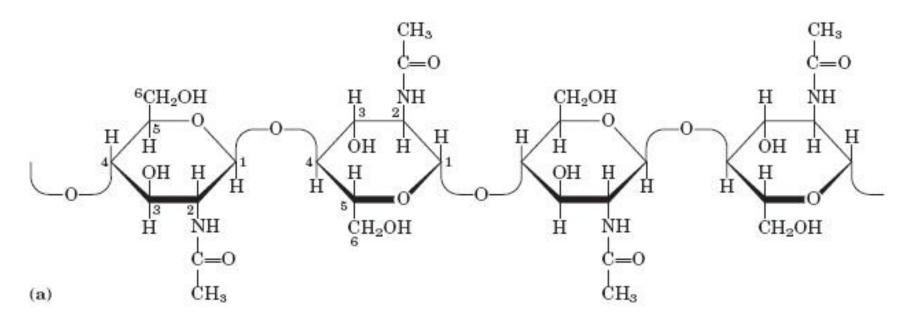
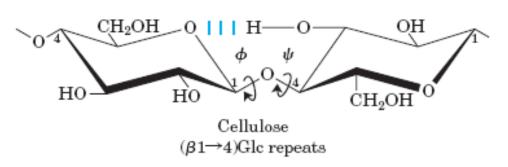
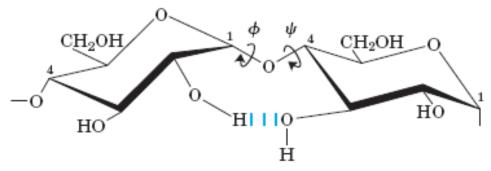


FIGURE 7-18 Chitin. (a) A short segment of chitin, a homopolymer of N-acetyl-p-glucosamine units in (β1→4) linkage. (b) A spotted June beetle (Pellidnota punetatia), showing its surface armor (exoskeleton) of chitin.

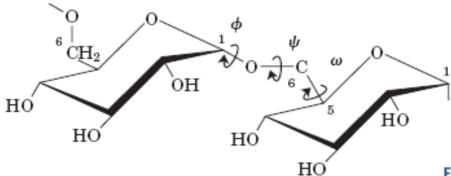


**(b)** 





Amylose  $(\alpha 1\rightarrow 4)$ Glc repeats



Dextran  $(\alpha 1 \rightarrow 6)$ Glc repeats, with  $(\alpha 1 \rightarrow 3)$  branches

**FIGURE 7-19** Conformation at the glycosidic bonds of cellulose, amylose, and dextran. The polymers are depicted as rigid pyranose rings joined by glycosidic bonds, with free rotation about these bonds. Note that in dextran there is also free rotation about the bond between C-5 and C-6 (torsion angle  $\omega$  (omega)).

### Bacterial and Algal Cell Walls Contain Structural Heteropolysaccharides

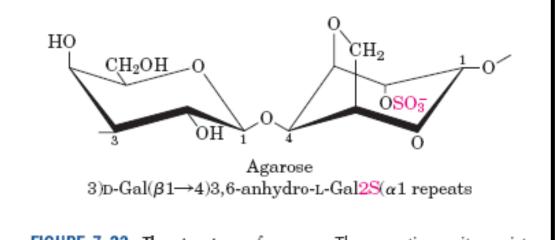


FIGURE 7-23 The structure of agarose. The repeating unit consists of p-galactose ( $\beta$ 1 $\rightarrow$ 4)-linked to 3,6-anhydro-μ-galactose (in which an ether ring connects C-3 and C-6). These units are joined by ( $\alpha$ 1 $\rightarrow$ 3) glycosidic links to form a polymer 600 to 700 residues long. A small fraction of the 3,6-anhydrogalactose residues have a sulfate ester at C-2 (as shown here).

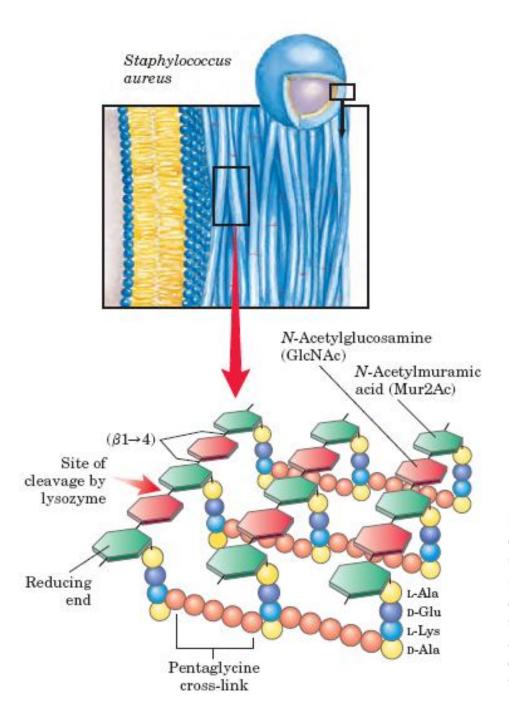
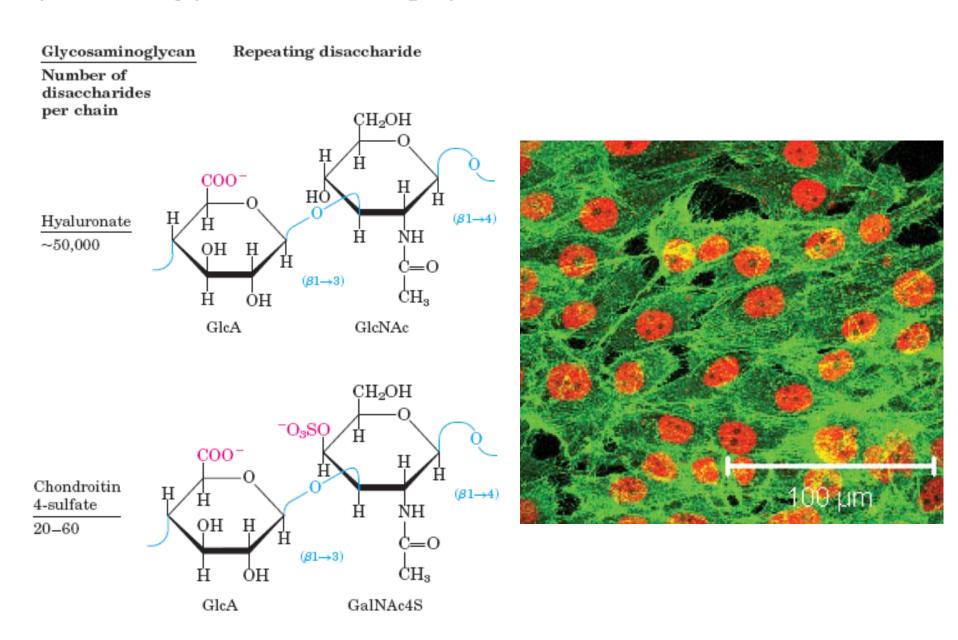


FIGURE 7-22 Peptidoglycan. Shown here is the peptidoglycan of the cell wall of *Staphylococcus aureus*, a gram-positive bacterium. Peptides (strings of colored spheres) covalently link *N*-acetylmuramic acid residues in neighboring polysaccharide chains. Note the mixture of ι and ρ amino acids in the peptides. Gram-positive bacteria have a pentaglycine chain in the cross-link. Gram-negative bacteria, such as *E. coli*, lack the pentaglycine; instead, the terminal ρ-Ala residue of one tetrapeptide is attached directly to a neighboring tetrapeptide through either ι-Lys or a lysine-like amino acid, diaminopimelic acid.

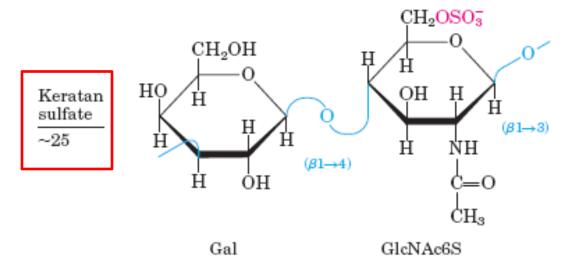
### Glycosaminoglycans Are Heteropolysaccharides of the Extracellular Matrix

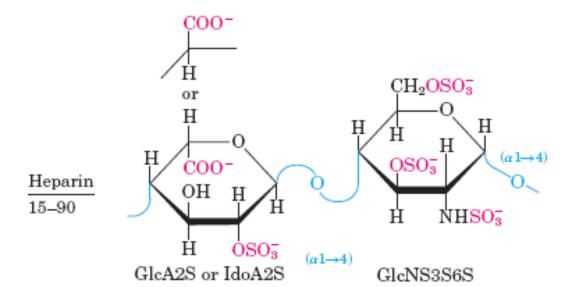


#### Glycosaminoglycan

Repeating disaccharide

Number of disaccharides per chain





### **Glycoconjugates:**

- **✓** Proteoglycans
- **✓** Glycoproteins
- **✓**Glycolipids
- **✓** Lipopolysaccharides

## نقش گلیکوکونژوگه ها

- □در میان کنش سلول با محیط خارج سلولی آن نقش دارد.
- □نشان دار کردن پروتئین ها برای انتقال به یک ارگانل یا مکان خاص در سلول و یا تخریب پروتئین های misfold
- □جایگاه شناسایی مولکول های پیام رسان خارج سلولی (مانند فاکتورهای رشد) یا انگل های خارج سلولی (باکتری یا ویروس)

هر سلول یوکاریوتی دارای یک پوشش الیگوساکاریدی به ضخامت چندنانومتر به نام Glycocalyx است (بخش های الیگوساکاریدی به غشای پلاسمایی سلول متصل هستند). این الیگوساکاریدها نقش های مهمی دارند:

- □ cell-cell recognition and adhesion,
- ☐ cell migration during development,
- ☐ blood clotting,
- ☐ immune response,
- ☐ wound healing,
- other cellular processes

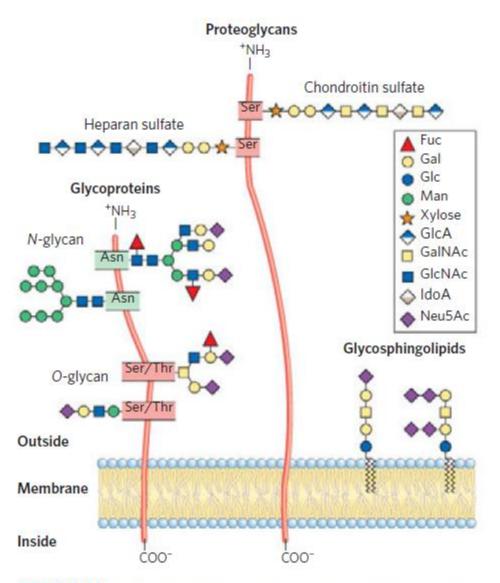


FIGURE 7–24 Glycoconjugates. The structures of some typical proteoglycans, glycoproteins, and glycosphingolipids described in the text.

Proteoglycans Are Glycosaminoglycan-Containing Macromolecules of the Cell Surface and Extracellular Matrix

# **Proteoglycans**

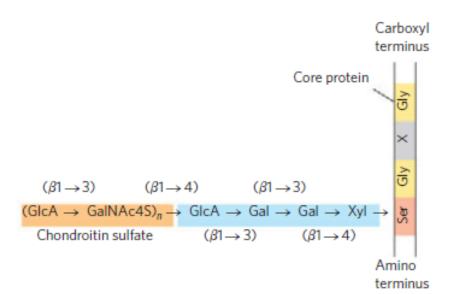
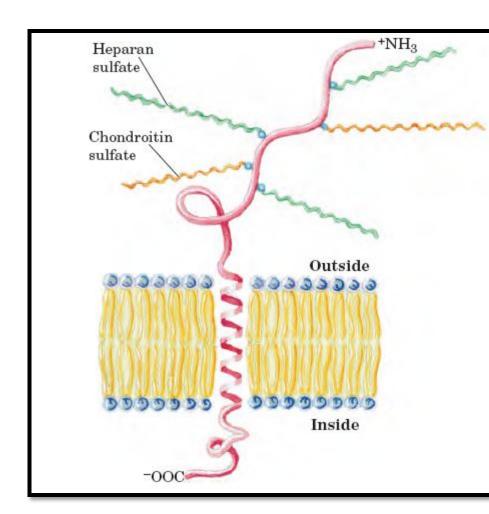


FIGURE 7–25 Proteoglycan structure, showing the tetrasaccharide bridge. A typical tetrasaccharide linker (blue) connects a glycosaminoglycan—in this case chondroitin 4-sulfate (orange)—to a Ser residue in the core protein. The xylose residue at the reducing end of the linker is joined by its anomeric carbon to the hydroxyl of the Ser residue.



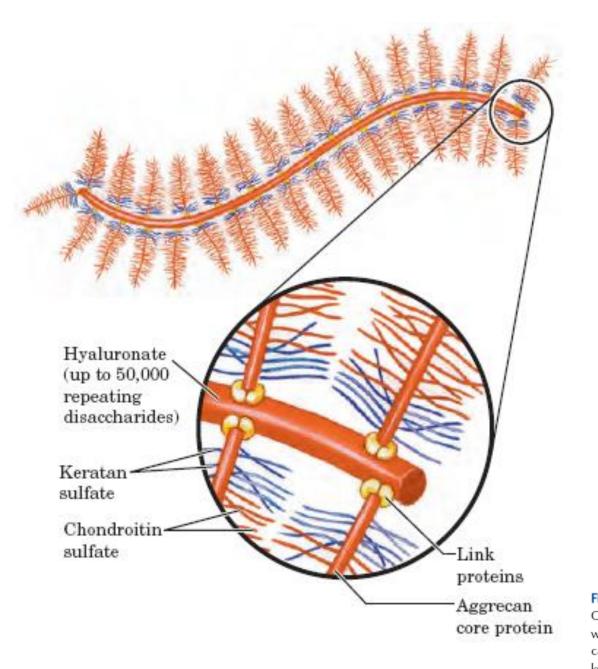
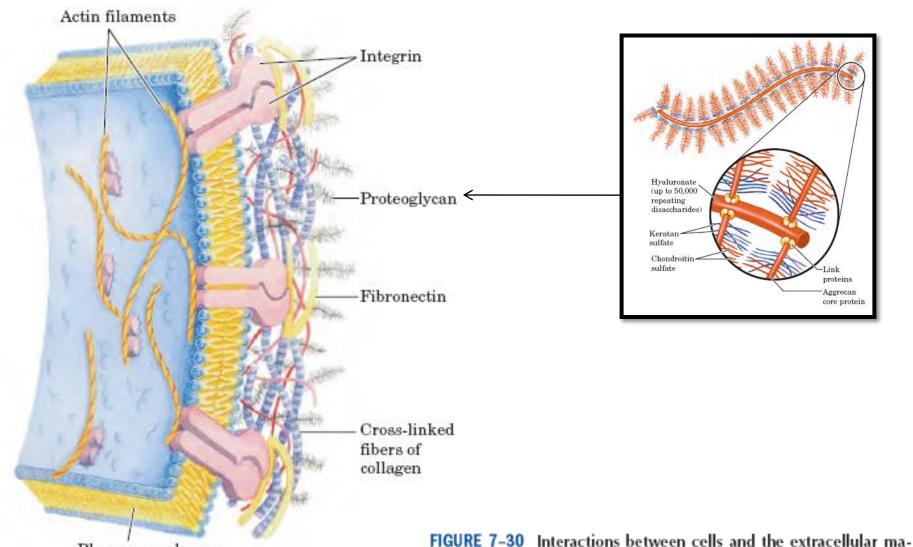


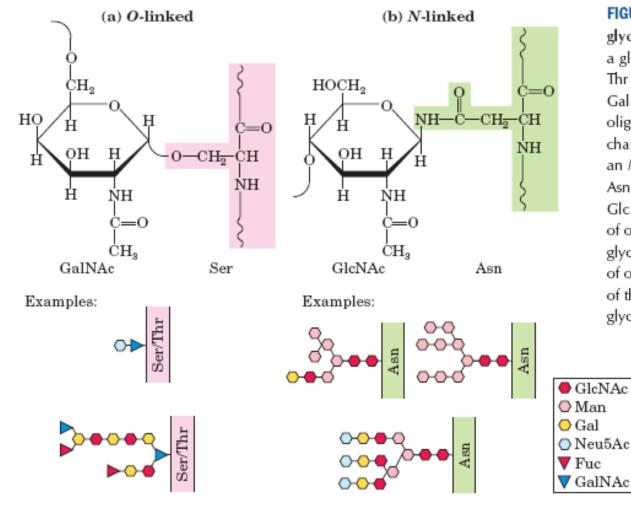
FIGURE 7-29 Proteoglycan aggregate of the extracellular matrix. One very long molecule of hyaluronate is associated noncovalently with about 100 molecules of the core protein aggrecan. Each aggrecan molecule contains many covalently bound chondroitin sulfate and keratan sulfate chains. Link proteins situated at the junction between each core protein and the hyaluronate backbone mediate the core protein–hyaluronate interaction.



Plasma membrane

trix. The association between cells and the proteoglycan of the extracellular matrix is mediated by a membrane protein (integrin) and by an extracellular protein (fibronectin in this example) with binding sites for both integrin and the proteoglycan. Note the close association of collagen fibers with the fibronectin and proteoglycan.

# **Glycoproteins**



**FIGURE 7-31** Oligosaccharide linkages in glycoproteins. (a) O-linked oligosaccharides have a glycosidic bond to the hydroxyl group of Ser or Thr residues (shaded pink), illustrated here with GalNAc as the sugar at the reducing end of the oligosaccharide. One simple chain and one complex chain are shown. (b) N-linked oligosaccharides have an N-glycosyl bond to the amide nitrogen of an Asn residue (shaded green), illustrated here with GlcNAc as the terminal sugar. Three common types of oligosaccharide chains that are N-linked in glycoproteins are shown. A complete description of oligosaccharide structure requires specification of the position and stereochemistry ( $\alpha$  or  $\beta$ ) of each glycosidic linkage.

# **Glycolipids**

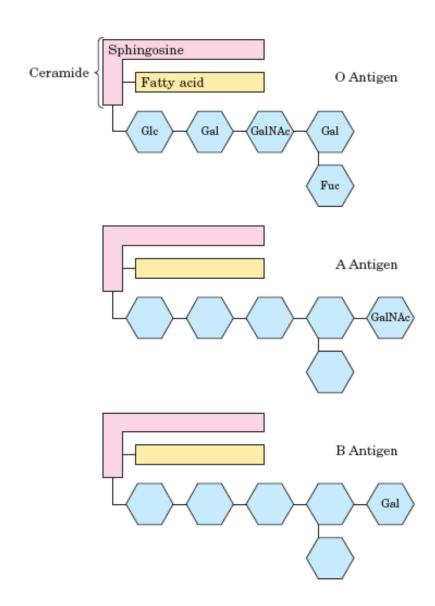


FIGURE 10-14 Glycosphingolipids as determinants of blood groups. The human blood groups (O, A, B) are determined in part by the oligosaccharide head groups (blue) of these glycosphingolipids. The same three oligosaccharides are also found attached to certain blood proteins of individuals of blood types O, A, and B, respectively. (Fuc represents the sugar fucose.)

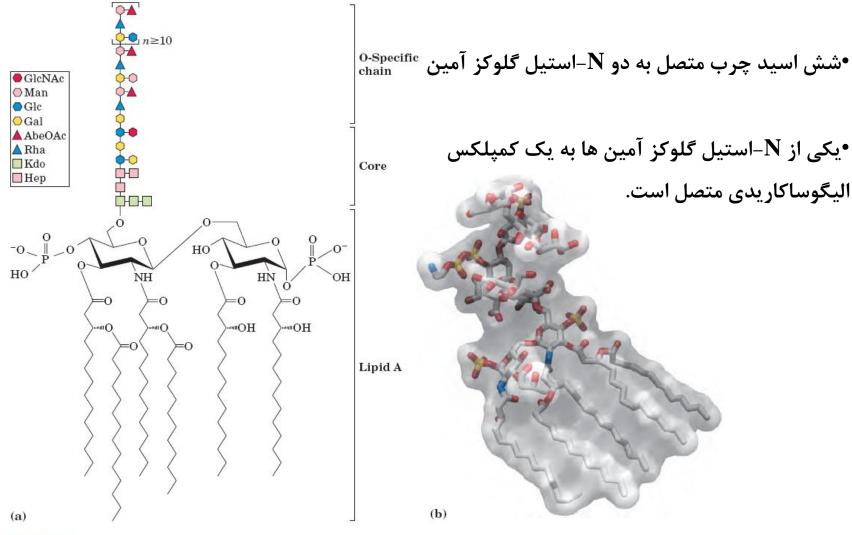


FIGURE 7-32 Bacterial lipopolysaccharides. (a) Schematic diagram of the lipopolysaccharide of the outer membrane of Salmonella typhimurium. Kdo is 3-deoxy-p-manno-octulosonic acid, previously called ketodeoxyoctonic acid; Hep is L-glycero-p-mannoheptose; AbeOAc is abequose (a 3,6-dideoxyhexose) acetylated on one of its hydroxyls. There are six fatty acids in the lipid A portion of the molecule. Different bacterial species have subtly different lipopolysaccharide structures, but they have in common a lipid region (lipid A), a core oligosaccharide, and an "O-specific" chain, which is the prin-

cipal determinant of the serotype (immunological reactivity) of the bacterium. The outer membranes of the gram-negative bacteria *S. ty-phimurium* and *E. coli* contain so many lipopolysaccharide molecules that the cell surface is virtually covered with O-specific chains. (b) The stick structure of the lipopolysaccharide of *E. coli* is visible through a transparent surface contour model of the molecule. The position of the sixth fatty acyl chain was not defined in the crystallographic study, so it is not shown.

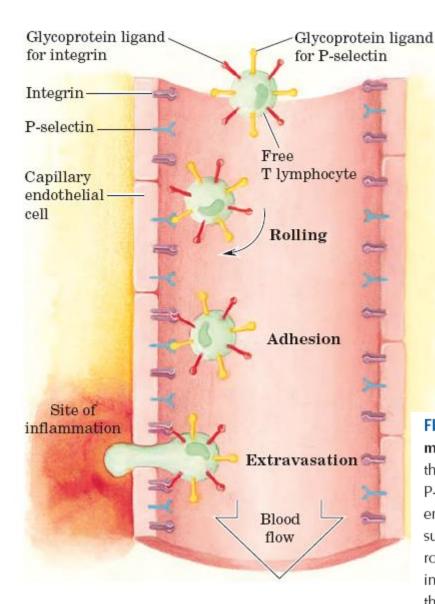


FIGURE 7-33 Role of lectin-ligand interactions in lymphocyte movement to the site of an infection or injury. A T lymphocyte circulating through a capillary is slowed by transient interactions between P-selectin molecules in the plasma membrane of the capillary endothelial cells and glycoprotein ligands for P-selectin on the T-cell surface. As it interacts with successive P-selectin molecules, the T cell rolls along the capillary surface. Near a site of inflammation, stronger interactions between integrin in the capillary surface and its ligand in the T-cell surface lead to tight adhesion. The T cell stops rolling and, under the influence of signals sent out from the site of inflammation, begins extravasation—escape through the capillary wall—as it moves toward the site of inflammation.

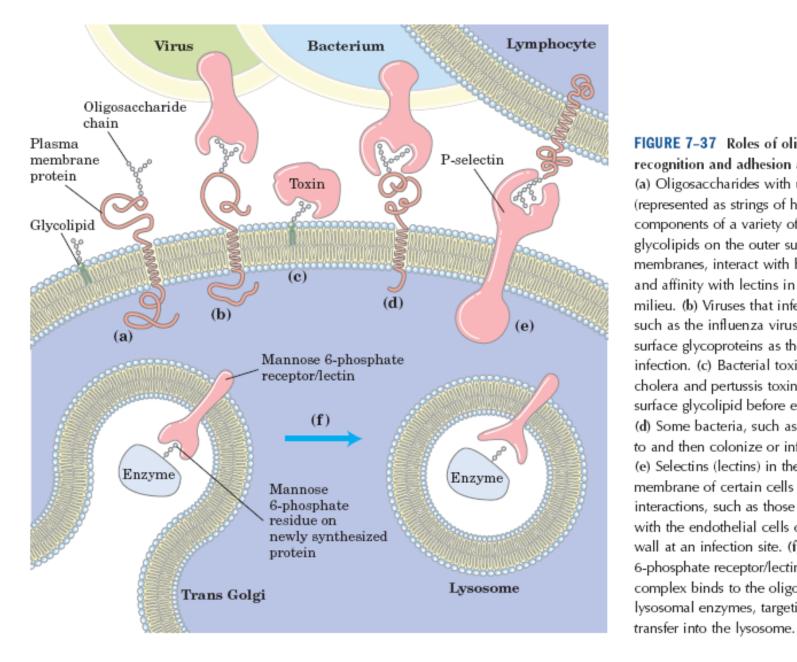


FIGURE 7-37 Roles of oligosaccharides in recognition and adhesion at the cell surface. (a) Oligosaccharides with unique structures (represented as strings of hexagons), components of a variety of glycoproteins or glycolipids on the outer surface of plasma membranes, interact with high specificity and affinity with lectins in the extracellular milieu. (b) Viruses that infect animal cells, such as the influenza virus, bind to cell surface glycoproteins as the first step in infection. (c) Bacterial toxins, such as the cholera and pertussis toxins, bind to a surface glycolipid before entering a cell. (d) Some bacteria, such as H. pylori, adhere to and then colonize or infect animal cells. (e) Selectins (lectins) in the plasma membrane of certain cells mediate cell-cell interactions, such as those of T lymphocytes with the endothelial cells of the capillary wall at an infection site. (f) The mannose 6-phosphate receptor/lectin of the trans Golgi complex binds to the oligosaccharide of lysosomal enzymes, targeting them for